

*Dedicated to Professor Emil Cordoş  
on the occasion of his 80<sup>th</sup> anniversary*

## HEALTH RISK ASSESSMENT ASSOCIATED WITH NITROGEN COMPOUNDS CONTAMINATED DRINKING WATER IN MEDIAS REGION

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**ABSTRACT.** High concentrations of nitrogen compounds in drinking water may cause negative health effects. The aim of the present study was to assess the content of nitrogen compounds namely, nitrates, nitrites and ammonium in drinking waters from Medias region (Medias, Copsa Mica towns and Tarnava village) and to investigate the health risk associated with the consumption of drinking water contaminated with these compounds. The health risk was calculated using chronic daily intake, hazard quotient and total hazard quotient. High concentrations of nitrite, nitrate and ammonium were found, at least one of the nitrogen compounds exceeding the maximum allowable concentrations (0.5 mg/L NO<sub>2</sub><sup>-</sup>, 0.5 mg/L NH<sub>4</sub><sup>+</sup>, 50 mg/L NO<sub>3</sub><sup>-</sup>) in about half of the analysed samples. Generally, the chronic daily intake values were lower for nitrite and ammonium than for nitrate. The hazard quotients for nitrate were higher than the critical unity value, indicating that the consumption of contaminated waters from Tarnava village and Medias town may cause potential non-carcinogenic risk. Moreover, for the same samples, the total hazard quotients were higher than the critical unity value, suggesting that potential adverse health effects may appear after the consumption of drinking water.

**Keywords:** *drinking water source, nitrite, nitrate, ammonium, chronic daily intake, hazard quotient, total hazard quotient*

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## INTRODUCTION

Ensuring access to adequate supply of safe drinking water is a major challenge all around the world. Groundwater is an important drinking water source, thus its quality protection is mandatory.

The presence of undesirable contaminants, such as nitrate ( $\text{NO}_3^-$ ), nitrite ( $\text{NO}_2^-$ ) or ammonium ( $\text{NH}_4^+$ ) may decrease the drinking waters quality by the unpleasant odour and taste and possible adverse health effects. The main exposure pathway to nitrogen compounds is the ingestion of water or foodstuffs that contain high levels of nitrates, nitrites and ammonium [1-5]. In the human body,  $\text{NH}_4^+$  is oxidised to  $\text{NO}_3^-$ , while  $\text{NO}_3^-$  under the action of specific enzymes is converted into  $\text{NO}_2^-$  that further reduces into N-nitroso compounds (nitrosamides, nitrosamines), with potential carcinogenic effects. Furthermore, the gastro intestinal tract and saliva are favourable environments for the conversion of nitrogenous compounds into  $\text{NO}_2^-$  and  $\text{NO}_3^-$  [6, 7, 8]. The  $\text{NO}_2^-$  reacts with the haemoglobin which is converted into methemoglobin, which stops carrying oxygen to all cell tissues [5, 9-10]. The most common effects that appear in case of bottled-fed infants are "blue-baby syndrome" or methemoglobinemia and developmental toxicity, which represent the main health concern regarding the  $\text{NO}_2^-$  and  $\text{NH}_4^+$  concentrations [2, 6, 7, 11]. Moreover the high concentrations of  $\text{NO}_3^-$  in the drinking water may increase the risk for bladder cancer [12].

Although the occurrence of  $\text{NO}_3^-$ ,  $\text{NO}_2^-$  and  $\text{NH}_4^+$  is part of the nitrogen cycle, the highest concentrations are generated by the anthropogenic activities. The agriculture (livestock, use of fertilizers) and the inadequate sewage management in industrial and household activities are considered the most important diffuse or non-point sources of  $\text{NO}_3^-$ ,  $\text{NO}_2^-$  and  $\text{NH}_4^+$  contamination [1-4, 8]. The main natural source of nitrogen compounds is the anaerobic organic material decomposition [8, 9, 13]. Several studies showed that the key source of high  $\text{NO}_3^-$ ,  $\text{NO}_2^-$  and  $\text{NH}_4^+$  levels in drinking-water are the sewage effluents runoff from household and agricultural activities [2, 13-16]. In Romania, high concentrations of  $\text{NO}_3^-$  and  $\text{NO}_2^-$  in drinking waters, which exceed the corresponding maximum allowable concentrations (MACs) were also reported [16-19]. Studies showed possible connection between the high concentrations of nitrogen compounds and the location of the well waters close to domestic and agricultural sources [19].

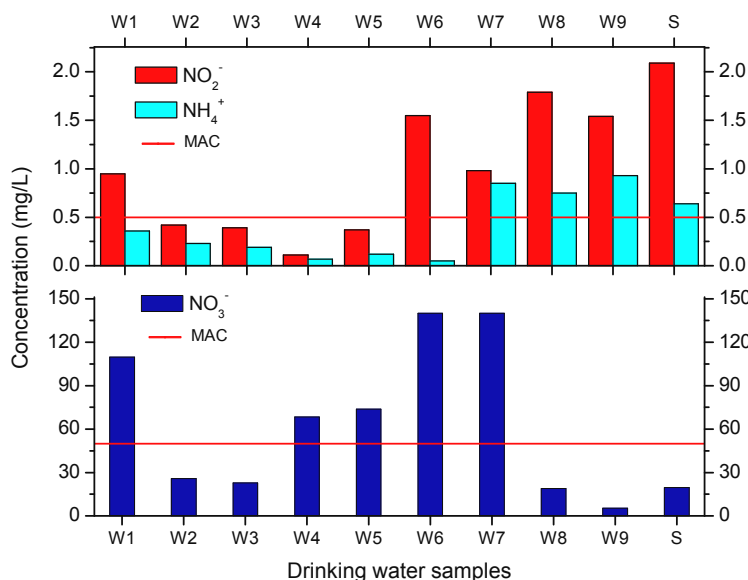
To prevent these health threats, European legislation and World Health Organization guidelines set maximum allowable concentrations for the  $\text{NO}_2^-$  (0.5 mg/L),  $\text{NH}_4^+$  (0.5 mg/L) and  $\text{NO}_3^-$  (50 mg/L) in drinking water [2, 20-22]. The health risks associated with contaminated drinking water consumption can be assessed using mathematical indices, such as chronic daily intake (CDI), hazard quotient (HQ) and total hazard quotient (THQ).

The aim of the current study was to assess the  $\text{NO}_3^-$ ,  $\text{NO}_2^-$  and  $\text{NH}_4^+$  levels in the drinking waters from Tarnava village, Copsa Mica and Medias towns. Furthermore, the obtained data was used to assess the health risk associated with the consumption of water contaminated with  $\text{NO}_3^-$ ,  $\text{NO}_2^-$  and  $\text{NH}_4^+$ . As the reference doses were established only for nitrate and nitrite, these two parameters were used to calculate the *HQ* and the *THQ*.

## RESULTS AND DISCUSSION

The results show high concentrations of  $\text{NO}_3^-$ ,  $\text{NO}_2^-$  and  $\text{NH}_4^+$  in the studied well and spring waters (figure 1).

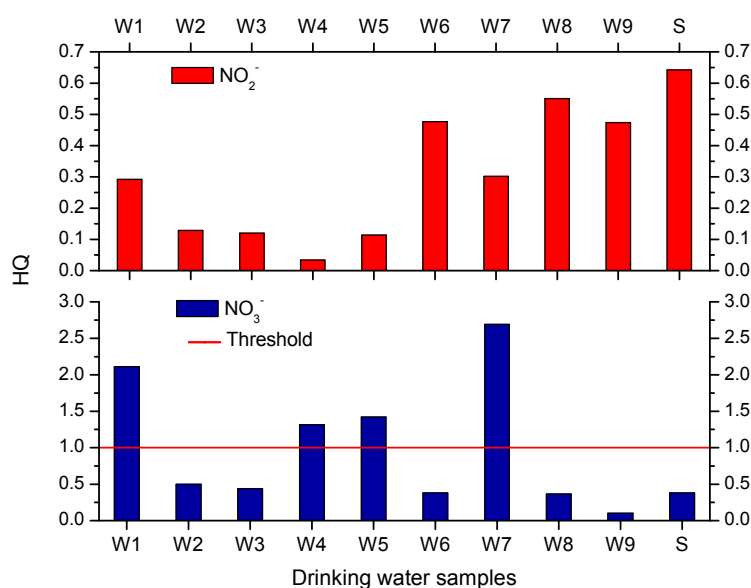
The  $\text{NO}_3^-$  concentrations in well water samples from Tarnava village (W6, W7) and Medias town (W1, W4, W 5) exceeded almost three times the *MAC* (50 mg/L), while the  $\text{NO}_2^-$  concentrations exceeded the *MAC* (0.5 mg/L) in water samples collected from all studied localities (W1, W6-W9, S) [16-18]. The measured  $\text{NO}_3^-$  concentration in spring sample S were lower than the *MAC*, while the  $\text{NO}_2^-$  concentration exceeded four times the *MAC* and the  $\text{NH}_4^+$  level is slightly exceeded [16-18]. Samples W7, W8 and W9 presented the highest  $\text{NH}_4^+$  concentrations, exceeding the *MAC* [16-18].



**Figure 1.** Concentrations of  $\text{NO}_2^-$ ,  $\text{NO}_3^-$  and  $\text{NH}_4^+$ , in drinking waters

The *HQ* and *CDI* approaches were used to assess the non-carcinogenic risks, specifically for methemoglobinemia, associated with the ingestion of  $\text{NH}_4^+$ ,  $\text{NO}_3^-$  and  $\text{NO}_2^-$  from drinking water sources. The *THQ* was used to summarize the total amount of chemicals ingested by drinking water. The *HQ* and *THQ* indices were applied only for  $\text{NO}_3^-$  and  $\text{NO}_2^-$ , since reference doses were established, by the IRIS (Integrated Risk Information System), U.S. E.P.A. (United States Environmental Protection Agency) for  $\text{NO}_2^-$  and  $\text{NO}_3^-$ , but not for  $\text{NH}_4^+$ .

Results indicate high values for the *HQ* for  $\text{NO}_3^-$ , exceeding the critical unity value in Tarnava village (W7) and Medias town (W1, W4, W5), which suggest potential non-carcinogenic effects for consumers [23, 24]. Rest of the studied well waters (W2, W3, W6, W8, W9) and spring (S) have *HQ* values below 1.00. The *HQ* results calculated for the  $\text{NO}_3^-$  concentrations range from 0.102 to 2.692 (Figure 2). Drinking water sources presented high  $\text{NO}_2^-$  levels, but low values for the *HQ*. The highest value were obtained for the spring sample (S) and the lowest value for W4 (Table 1). Possible sources that can contribute to the high concentrations  $\text{NO}_2^-$  and  $\text{NO}_3^-$  are the agricultural and the domestic activities.

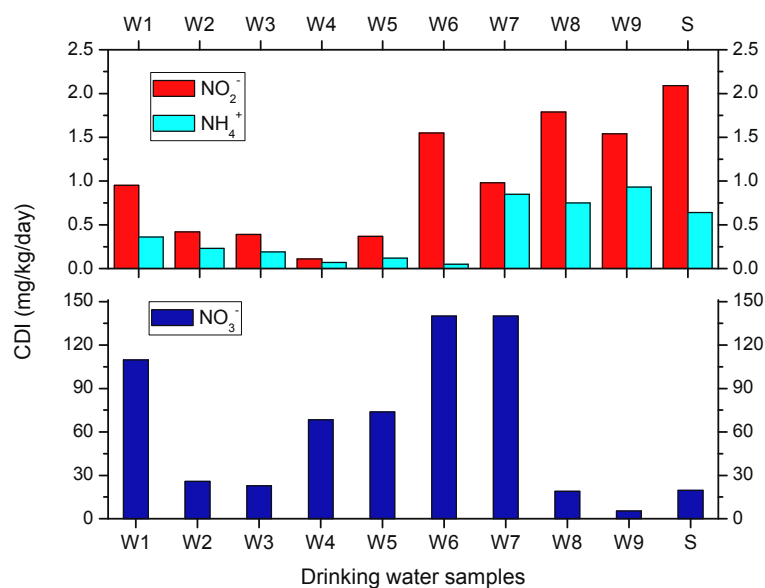


**Figure 2.** The hazard quotients for  $\text{NO}_3^-$  and  $\text{NO}_2^-$  in drinking waters

**Table 1.** The *CDI*, *HQ* and *THQ* for the drinking waters

	W1		W2		W3		W4		W5	
	<i>CDI</i> mg/kg/day	<i>HQ</i>	<i>CDI</i> mg/kg/day	<i>HQ</i>	<i>CDI</i> mg/kg/day	<i>HQ</i>	<i>CDI</i> mg/kg/day	<i>HQ</i>	<i>CDI</i> mg/kg/day	<i>HQ</i>
$\text{NO}_2^-$	0.029	0.292	0.013	0.129	0.012	0.120	0.003	0.034	0.011	0.114
$\text{NO}_3^-$	3.378	2.112	0.797	0.498	0.701	0.438	2.101	1.313	2.270	1.419
$\text{NH}_4^+$	0.011		0.007		0.006		0.002		0.004	
<i>THQ</i>	2.404		0.628		0.558		1.347		1.533	
	S		W6		W7		W8		W9	
	<i>CDI</i> mg/kg/day	<i>HQ</i>	<i>CDI</i> mg/kg/day	<i>HQ</i>	<i>CDI</i> mg/kg/day	<i>HQ</i>	<i>CDI</i> mg/kg/day	<i>HQ</i>	<i>CDI</i> mg/kg/day	<i>HQ</i>
$\text{NO}_2^-$	0.064	0.643	0.048	0.477	0.030	0.302	0.055	0.551	0.047	0.474
$\text{NO}_3^-$	0.606	0.379	4.308	2.692	4.308	2.692	0.585	0.365	0.163	0.102
$\text{NH}_4^+$	0.020		0.002		0.026		0.023		0.029	
<i>THQ</i>	1.022		3.169		2.994		0.916		0.576	

The *CDI* for  $\text{NO}_3^-$  and  $\text{NO}_2^-$  showed low values for samples from Copsa Mica and Medias towns (Figure 3) and high values for water samples collected from Tarnava village. The *CDI* ranged between 0.003 and 0.064 mg/kg/day for  $\text{NO}_2^-$  and 0.163 and 4.308 mg/kg/day for  $\text{NO}_3^-$  with a mean of 0.031 mg/kg/day and 1.922 mg/kg/day, respectively.

**Figure 3.** The chronic daily intake for  $\text{NO}_3^-$ ,  $\text{NO}_2^-$  and  $\text{NH}_4^+$  in drinking waters

The *CDI* calculated for the samples from Tarnava village (W6, W7) and Medias town (W1, W4, W5) exceeds 1.00. Obtained *CDI* values for  $\text{NH}_4^+$  ranged from 0.002 to 0.029 mg/kg/day, with a mean of 0.013 mg/kg/day. The lowest values were obtained for W4 and W6, while the highest for W9. According to Buss et al. (2014),  $\text{NH}_4^+$  is typically present in the wastewater discharges and landfill leachates, which could represent a possible  $\text{NH}_4^+$  source for well waters [25]. Low *CDI* values for  $\text{NO}_2^-$  and  $\text{NO}_3^-$  ( $< 0.2$  mg/kg/day) and *HQ* values for  $\text{NO}_2^-$  and  $\text{NO}_3^-$  ( $< 0.2$ ) were found in river water samples used as drinking water sources in Poland [25]. Respective values give no cause for concern regarding the non-carcinogenic risk at  $\text{NO}_3^-$  and  $\text{NO}_2^-$  [26].

The *THQ* values were higher than 1.00 for the spring sample S and for the well waters from Tarnava village (W6, W7) and Medias (W1, W4, W5). In the case of well waters from Copsa Mica (W8, W9) and from Medias town (W2, W3) the *THQ* values were lower than 1.00 (Table 1).

## CONCLUSIONS

The studied well water samples from Tarnava village are contaminated with  $\text{NO}_3^-$ ,  $\text{NO}_2^-$  and  $\text{NH}_4^+$ . The *HQ* for  $\text{NO}_3^-$  indicated values above the critical unity value in case of one water sample, while the *HQ* for  $\text{NO}_2^-$  were below the critical unity value. Obtained *THQ* values for the two well water samples from Tarnava village were 3.169 and 2.994. Chronic daily intake of  $\text{NO}_2^-$  and  $\text{NH}_4^+$  were below 0.100 mg/kg/day, while the *CDI* of  $\text{NO}_3^-$  was 4.308 mg/kg/day.

Well water and spring samples from Medias town are characterized by low  $\text{NO}_2^-$  and  $\text{NH}_4^+$  concentrations, except one sample, which exceeds the *MAC*. Three water samples exceeded the *MAC* for  $\text{NO}_3^-$ . Three waters from Medias town showed possible non-charcinogenic risk, according to the *HQ* values, which give a cause of concern. While *HQ* calculated for  $\text{NO}_3^-$  showed values below 1.00. Results for the *THQ* ranged between 0.558 and 2.404. The obtained values for the *CDI* of  $\text{NO}_2^-$  ranged between 0.003 and 0.029 mg/kg/day, the calculated values for the *CDI* of  $\text{NO}_3^-$  ranged from 0.606 to 3.378 mg/kg/day, while the *CDI* of  $\text{NH}_4^+$  ranged from 0.002 to 0.011 mg/kg/day.

The  $\text{NO}_3^-$  concentrations measured for well water samples collected from Copsa Mica town are lower than *MAC*. While the  $\text{NO}_2^-$  and the  $\text{NH}_4^+$  exceeded the *MACs*. The *HQ* for  $\text{NO}_2^-$  and  $\text{NO}_3^-$  were lower as the critical unity value, indicating that the water samples present no potential non-carcinogenic

risks. The *THQ* results were below 1.00, suggesting that potential adverse health effects may not appear after the consumption of waters from the studied water sources. Chronic daily intake for  $\text{NO}_2^-$  and  $\text{NO}_3^-$  were below 0.200 mg/kg/day, while for  $\text{NH}_4^+$  below 0.100. The domestic and agricultural activities were taken into consideration as possible sources for the high nitrogen compounds concentrations. Under these circumstances it is recommended a filtration process for the drinking water before consumption and if it is possible a disinfection of the well waters, for the consumer which have no alternative drinking water sources.

## EXPERIMENTAL SECTION

### Study area, sampling and chemical analysis

The study region is part of Transylvania, localized in central Romania (Figure 4). This part of the country is characterized by an average annual temperature of 8.6 °C and an annual precipitation range of 700 mm [24]. Population from rural and small urban areas in the study area (Tarnava village, Copsa Mica and Medias towns) practice agricultural activities as livestock growing and crop cultivation. Private well waters and public natural springs are used as drinking water sources [17, 18].

One spring sample (S1) from Medias, two well water samples (W1, W2) from Tarnava village and two well water samples (W3, W4) from Copsa Mica were collected [17, 18]. The drinking water samples were collected during summer of 2015, one from each sampling point (Figure 4) in polyethylene bottles and kept at 4 °C in a refrigerator until the chemical analysis. Water samples were filtered using cellulose acetate membrane filters with pore-size of 0.45  $\mu\text{m}$  [17, 18]. The  $\text{NO}_2^-$ ,  $\text{NO}_3^-$  and  $\text{NH}_4^+$  concentrations were measured by ion-liquid chromatography, according to ISO 10304-1:2007, using the 761 Compact IC (Methrom, Herisau, Switzerland) and the  $\text{NH}_4^+$  as indophenol blue complex, according to SR ISO 7150-1:2001 by spectrophotometer Lambda 25, (Perkin-Elmer, Beaconsfield, UK).

The accuracy of  $\text{NO}_2^-$ ,  $\text{NO}_3^-$  and  $\text{NH}_4^+$  determinations was tested by analysing Nitrite standard solution (40 mg/L), Nitrate standard solution (15.0 mg/L), and Ammonium standard solution (1.0 mg/L) purchased from Merck. The found results were in good agreement with the certified values for all parameters. The recovery expressed as relative standard deviation degree ranged between 89 % and 100 %.

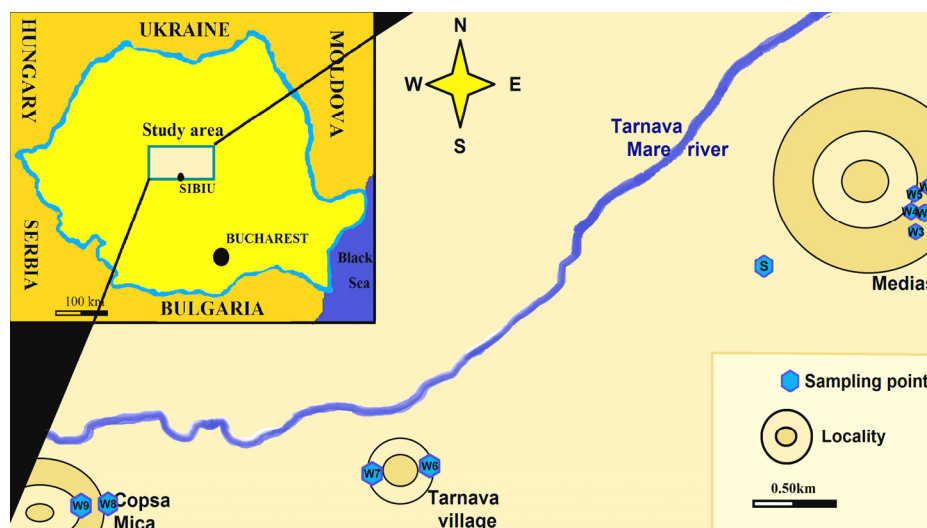


Figure 4. Study area and the location of the sampling points

#### Human health risk assessment

In order to estimate health risk associated with  $\text{NO}_2^-$  and  $\text{NO}_3^-$  in drinking water, the *CDI* was calculated using the following equation [27]:

$$CDI = \frac{C \times DI}{BW} \quad (1)$$

Where *C* represents the  $\text{NO}_3^-$  and  $\text{NO}_2^-$  concentrations in drinking water (mg/L), *DI* is the daily intake rate (2 l/day) and *BW* represents the body weight (72 kg) [27].

The hazard quotient (*HQ*) estimates the non-carcinogenic risk posed by  $\text{NO}_3^-$  and  $\text{NO}_2^-$  in drinking water, and was calculated using the following equation [27]:

$$HQ = \frac{CDI}{RfD} \quad (2)$$

The *RfD* represents the reference dose with values of 0.10 mg/kg/day for  $\text{NO}_2^-$  and 1.6 mg/kg/day for  $\text{NO}_3^-$ . The reference doses were set by EPA's IRIS (Integrated Risk Information System) Program [28, 29]. There are no potential non-carcinogenic risks for the exposed population in cases when the *HQ* does not exceed unity ( $HQ < 1.00$ ) [27, 29]. In the current study the oral exposure was taken into consideration for the count of the *HQ*.

*THQ* represents a summation of the ingestion of  $\text{NO}_3^-$  and  $\text{NO}_2^-$  through drinking water [30].



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